. .

216 parametersAtomic scattering factors
$$w = 1/[\sigma^2(F_o^2) + (0.1214P)^2$$
from International Tables $+ 0.0307P]$ from Crystallography (1992,where $P = (F_o^2 + 2F_c^2)/3$ Vol. C, Tables 4.2.6.8 and $(\Delta/\sigma)_{max} = -0.094$ 6.1.1.4) $\Delta \rho_{max} = 0.290$ e Å⁻³Absolute configuration: $\Delta \rho_{min} = -0.237$ e Å⁻³Flack (1983)

 Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$U_{\rm eq} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_i^* \mathbf{a}_i \cdot \mathbf{a}_j.$$

	x	у	Z	U_{eq}
S	0.92832 (10)	0.51629 (9)	0.86263 (3)	0.0538 (3)
01	0.8354 (5)	0.6597 (3)	0.88601 (11)	0.0825 (9)
02	0.8600 (2)	0.3954 (3)	0.70150 (8)	0.0474 (5)
03	0.6470 (3)	0.5830 (3)	0.68440 (9)	0.0577 (6)
CI	0.8600 (4)	0.3407 (4)	0.90273 (12)	0.0455 (6)
C2	0.7631 (4)	0.3635 (4)	0.95041 (13)	0.0546 (7)
C3	0.7202 (5)	0.2284 (5)	0.98276 (13)	0.0588 (8)
C4	0.7738 (4)	0.0711 (4)	0.96915 (13)	0.0565 (8)
C5	0.8691 (5)	0.0501 (4)	0.92021 (14)	0.0548 (7)
C6	0.9125 (4)	0.1838 (4)	0.88756 (12)	0.0494 (7)
C7	0.7331 (7)	-0.0760 (6)	1.00576 (2)	0.0862 (13)
C8	0.8320 (3)	0.4630 (3)	0.79788 (11)	0.0420 (6)
C9	0.6446 (4)	0.4595 (4)	0.79664 (12)	0.0508 (7)
C10	0.5887 (4)	0.3595 (4)	0.74638 (14)	0.0522 (7)
C11	0.6811 (4)	0.4146 (4)	0.69368 (12)	0.0449 (6)
C12	0.9264 (3)	0.4328 (3)	0.75319(11)	0.0407 (6)
C13	0.6377 (5)	0.3249 (4)	0.6392 (2)	0.0625 (8)
C14	0.6653 (6)	0.4585 (7)	0.5951 (2)	0.0785 (11)
C15	0.6204 (6)	0.6126 (5)	0.6258 (2)	0.0750 (11)
C16	1.11562 (4)	0.4248 (4)	0.74984 (15)	0.0585 (8)

Table 2. Selected geometric parameters (Å, °)

S-01	1.486 (3)	C8—C9	1.493 (4)	
SC8	1.773 (3)	C9-C10	1.511 (5)	
S-C1	1.798 (3)	C10-C11	1.520 (4)	
C1C2	1.383 (5)	C11—O3	1.410 (4)	
C1C6	1.387 (4)	C11—O2	1.446 (4)	
C2-C3	1.382 (5)	C11—C13	1.525 (5)	
C3C4	1.383 (5)	O2—C12	1.371 (3)	
C4C5	1.399 (5)	C12—C16	1.512 (4)	
C4—C7	1.512 (5)	C13-C14	1.524 (6)	
C5—C6	1.378 (4)	C14C15	1.490 (6)	
C8-C12	1.324 (4)	C15—O3	1.429 (4)	
O1—S—C8	107.4 (2)	C8-C9-C10	108.7 (2)	
01—S—C1	105.7 (2)	C9-C10-C11	110.5 (2)	
C8SC1	97.79 (12)	O3-C11O2	108.3 (2)	
C2C1C6	120.3 (3)	O3—C11—C10	108.7 (3)	
C2-C1S	119.8 (2)	O2—C11—C10	109.9 (2)	
C6C1S	119.8 (2)	O3-C11-C13	106.6 (2)	
C3C2C1	119.2 (3)	O2-C11-C13	106.3 (3)	
C2-C3-C4	121.7 (3)	C10-C11-C13	116.7 (3)	
C3-C4-C5	118.3 (3)	C12-02-C11	118.1 (2)	
C3-C4-C7	121.8 (3)	C8-C12-O2	122.7 (2)	
C5-C4-C7	119.9 (3)	C8-C12-C16	128.1 (3)	
C6-C5-C4	120.5 (3)	O2-C12-C16	109.2 (2)	
C5-C6-C1	120.0 (3)	C14C13C11	102.2 (3)	
C12-C8-C9	123.3 (3)	C15-C14-C13	103.0 (3)	
C12-C8-S	119.7 (2)	O3-C15-C14	107.5 (3)	
C9—C8—S	117.0 (2)	C11—O3—C15	110.1 (3)	

The determination of the absolute configuration was possible from the known S configuration of the sulfoxide group in the starting heterodiene; using the method described by Flack (1983), the absolute configuration was confirmed by the calculations (*SHELXL93*; Sheldrick, 1993). H atoms were refined as rigid groups using the *AFIX* card of the *SHELXL93* program.

Data collection: DIF4 (Stoe & Cie, 1988a). Cell refinement: DIF4. Data reduction: REDU4 (Stoe & Cie, 1988b). Pro-

gram(s) used to solve structure: *SHELXS*86 (Sheldrick, 1990) option *TREF*. Program(s) used to refine structure: *SHELXL*93. Molecular graphics: *ORTEP* (Johnson, 1965).

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Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: PA1118). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Boger, D. L. & Weinzed, S. M. (1987). Hetero-Diels-Alder Methodology in Organic Synthesis. New York: Academic Press.
- Bonfand, E., Gosselin, P. & Maignan, C. (1992). Tetrahedron Lett. 33, 2347-2348.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
- Sheldrick, G. M. (1990). SHELXS86. Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. Univ. of Göttingen, Germany.
- Stoe & Cie (1988a). DIF4. Diffractometer Control Program. Version 6.2. Stoe & Cie, Darmstadt, Germany.
- Stoe & Cie (1988b). *REDU4. Data Reduction Program.* Version 6.2. Stoe & Cie, Darmstadt, Germany.

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The Diels-Alder Adduct of an Enantiopure 2-Sulfinyldiene and Maleimide: (1*S*,2*R*,3*S*,*SR*)-3-Methyl-5-*p*-toluenesulfinylcyclohex-4-en-1,2dicarboximide

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(Received 23 March 1994; accepted 24 June 1994)

Abstract

The title compound, $C_{16}H_{17}NO_3S$, is the single adduct obtained by cycloaddition of (E)-(+)-(R)-2-*p*-toluenesulfinyl-1,3-pentadiene with maleimide. The ab-

solute configuration of the stereocentres is established and some mechanistic aspects are deduced.

Comment

Control over asymmetric induction in Diels-Alder cycloadditions is a matter of continuing interest (Taschner, 1989). We reported recently that enantiopure 2-sulfinyldiene may readily be synthesized from (SR)vinylsulfoxide (Bonfand, Gosselin & Maignan, 1993). We then examined the cycloaddition of (E)-(+)-(R)-2*p*-toluenesulfinyl-1,3-pentadiene in order to study both endo/exo and diastereofacial selectivities. This diene was reacted with maleimide in ether at room temperature and the reaction was complete after 10 h. Only one of four possible stereoisomers, (I), was isolated.



The configuration of the three contiguous stereocentres was shown to be 1S,2R,3S (C12, C11 and C10, respectively) by single-crystal X-ray diffraction. The title compound results from an exclusive endo approach by a completely diastereofacial selective route, *i.e.* from the less hindered side of the diene, which is also the most nucleophilic. This indicates an s-trans conformation of the S=O and C=C bonds in the diene, at least in the transition state.



Fig. 1 ORTEP plot of C16H17NO3S. For the sake of clarity the displacement parameters of the H atoms have been divided by ten. Displacement ellipsoids are plotted at the 50% probability level.

Experimental

Recrystallization from CH2Cl2/Et2O afforded the title compound as colourless platelets and a parallelepipedic crystal

was chosen for X-ray analysis. Its quality was tested using Laue photographs.

Crystal data

$C_{16}H_{17}NO_3S$	Mo $K\alpha$ radiation
$M_r = 303.37$	$\lambda = 0.71069 \text{ Å}$
Orthorhombic	Cell parameters from 36
$P2_{1}2_{1}2_{1}$	reflections
a = 8.5143 (7) Å	$\theta = 12.75 - 15.7^{\circ}$
<i>b</i> = 11.8171 (6) Å	$\mu = 0.217 \text{ mm}^{-1}$
c = 14.7520 (9) Å	T = 293 (2) K
V = 1484.3 (2) Å ³	Parallelepiped
Z = 4	$0.608 \times 0.532 \times 0.456 \text{ mm}$
$D_x = 1.358 \text{ Mg m}^{-3}$	Colourless

frequency: 60 min intensity variation: 1.3%

Data collection

Stoe Siemens AED-2 dif-	$R_{\rm int} = 0.0342$
fractometer	$\theta_{\rm max} = 30^{\circ}$
ω -2 θ scans	$h = -11 \rightarrow 11$
Absorption correction:	$k = -16 \rightarrow 16$
none	$l = -20 \rightarrow 20$
4922 measured reflections	3 standard reflections
4328 independent reflections	frequency: 60 min
3727 observed reflections	intensity variation:
$[I > 2\sigma(I)]$	

Refinement

N 01

02 03 C1 C2 C3 C4 C5

C6

C7 C8

C9 C10 C11 C12

C13 C14

C15

C16

$\Delta \rho_{\rm max} = 0.248 \ {\rm e} \ {\rm A}^{-3}$
$\Delta \rho_{\rm min} = -0.219 \ {\rm e} \ {\rm \AA}^{-3}$
Extinction correction: none
Atomic scattering factors
from International Tables
for Crystallography (1992,
Vol. C, Tables 4.2.6.8 and
6.1.1.4)
Absolute configuration:
Flack (1983)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$U_{\rm eq} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$			
x	у	z	U_{eq}
0.14421 (5)	0.91874 (4)	0.81601 (3)	0.04102 (12)
0.3896 (2)	0.88709 (14)	0.53164 (10)	0.0444 (4)
0.2347 (2)	0.97751 (14)	0.88926 (9)	0.0568 (4)
0.5133 (3)	1.02400 (13)	0.61450 (13)	0.0700 (5)
0.2948 (3)	0.7207 (2)	0.47557 (12)	0.0765 (6)
0.1252 (2)	1.01846 (13)	0.72563 (10)	0.0356 (3)
0.1936 (2)	1.1233 (2)	0.73248 (13)	0.0414 (4)
0.1756 (2)	1.2012 (2)	0.6635 (2)	0.0493 (4)
0.0905 (3)	1.1753 (2)	0.58651 (13)	0.0581 (6)
0.0203 (5)	1.0698 (3)	0.5822 (2)	0.0845 (10)
0.0356 (4)	0.9913 (2)	0.6512 (2)	0.0670(7)
0.0754 (5)	1.2613 (3)	0.5107 (2)	0.0887 (11)
0.2774 (2)	0.82421 (13)	0.76250 (10)	0.0325 (3)
0.2239 (2)	0.72916 (14)	0.72617 (12)	0.0374 (3)
0.3391 (2)	0.65245 (12)	0.67886 (11)	0.0349 (3)
0.4468 (2)	0.72360 (13)	0.61668 (10)	0.0308 (3)
0.5161 (2)	0.82832 (14)	0.66435 (9)	0.0314 (3)
0.4487 (2)	0.85171 (15)	0.75922 (11)	0.0363 (3)
0.2588 (3)	0.5523 (2)	0.6320 (2)	0.0548 (5)
0.3651 (2)	0.7725 (2)	0.53340 (11)	0.0422 (4)
0.4769 (2)	0.9272 (2)	0.60267 (11)	0.0408 (3)

Table 2. Selected geometric parameters (Å, °)

	0	1	, ,
S—01	1.498 (2)	C9-C10+	1.507 (2)
SC8	1.7766 (15)	C10-C14	1.532 (3)
SC1	1.787 (2)	C10-C11	1.545 (2)
C1—C2	1.373 (2)	CI1C15	1.525 (2)
C1C6	1.375 (3)	C11—C12	1.541 (2)
C2C3	1.380 (3)	C12C16	1.518 (2)
C3C4	1.382 (3)	C12C13	1.538 (2)
C4—C5	1.384 (5)	C15-03	1.209 (2)
C4C7	1.517 (3)	C15—N	1.370 (3)
C5C6	1.383 (4).	C16—O2	1.197 (3)
C8C9	1.325 (2)	C16N	1.370 (2)
C8C13	1.495 (2)		
O1SC8	106.49 (9)	C9-C10-C14	112.5 (2)
01 S C1	106.21 (8)	C9-C10-C11	109.49 (12)
C8—S—C1	98.10 (7)	C14-C10-C11	114.68 (15)
C2C1C6	120.3 (2)	C15-C11-C12	103.77 (13)
C2C1S	120.13 (13)	C15-C11-C10	114.44 (14)
C6C1S	119.5 (2)	C12C11C10	113.17 (11)
C1C2C3	120.0 (2)	C16-C12-C13	108.95 (14)
C4—C3—C2	121.1 (2)	C16-C12-C11	105.10 (12)
C3C4C5	117.7 (2)	C13-C12-C11	114.61 (12)
C3C4 C7	120.1 (3)	C8C13C12	110.78 (13)
C5C4C7	122.2 (3)	O3C15N	124.2 (2)
C6C5C4	121.9 (2)	O3C15C11	127.0 (2)
C1C6C5	118.9 (2)	NC15C11	108.68 (15)
C9C8C13	120.42 (14)	02C16N	125.6 (2)
C9—C8—S	119.56 (13)	O2C16C12	126.2 (2)
C13-C8-S	120.02 (12)	NC16C12	108.1 (2)
C8-C9-C10	118.26 (14)	C16NC15	114.23 (15)

The determination of the absolute configuration was possible from the known R configuration of the sulfoxide group in the starting diene; using the method described by Flack (1983), tha absolute configuration was confirmed by the calculations [*SHELXL93* (Sheldrick, 1993) option applied on non-centrosymmetric space groups]. H atoms were refined as rigid groups using the *AFIX* card of the *SHELXS*93 program.

Data collection: *DIF4* (Stoe & Cie, 1988a). Cell refinement: *DIF4*. Data reduction: *REDU4* (Stoe & Cie, 1988b). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990), option *TREF*. Program(s) used to refine structure: *SHELXL93*. Molecular graphics: *ORTEP* (Johnson, 1965).

Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: PA1119). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Bonfand, E., Gosselin, P. & Maignan, C. (1993). Tetrahedron Asymmetry, 4, 1667-1676.
- Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. Univ. of Göttingen, Germany.
- Stoe & Cie (1988a). DIF4. Diffractometer Control Program. Version 6.2. Stoe & Cie, Darmstadt, Germany.
- Stoe & Cie (1988b). REDU4. Data Reduction Program. Version 6.2. Stoe & Cie, Darmstadt, Germany.
- Taschner, M. J. (1989). Asymmetric Diels-Alder Reaction. Organic Synthesis Theory and Application, edited by T. Hudlicky, pp. 1– 101. Greenwich, CT: Jai Press.

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5-Ethyl-2'-deoxycytidine, C₁₁H₁₇N₃O₄

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(Received 18 October 1993; accepted 18 May 1994)

Abstract

The deoxyribose ring in the title compound adopts the twist conformation $({}_{3}^{2}T)$ with a glycosyl torsion angle of 231.0 (6)°. The pseudo-rotation parameters are $P = 180.29 (5)^{\circ}$ and $\tau_{m} = 38.3 (5)^{\circ}$. The exocyclic side chain at C5' has the t conformation $[\gamma = 174.8 (11)^{\circ}]$. The ethyl group at C5 is on the same side of the pyrimidine plane as the O4' atom of the furanose ring.

Comment

5-Ethyl-2'-deoxyuridine (EtdUrd) has been shown to be a relatively potent inhibitor of herpes simplex virus (types 1 and 2) replication (De Clerq & Rosenwirth, 1985; De Clerq & Shugar, 1975; Schinazi, Scott, Peters, Rice & Nahmias, 1985). 5-Ethyl-2'-deoxycytidine (EtdCyd) was prepared as a potential antiviral agent, based on the potency of EtdUrd (De Clerq & Shugar, 1975; Kulikowski & Shugar, 1974).



A perspective ORTEPII (Johnson, 1976) drawing of the molecule is shown in Fig. 1. The bond lengths and angles of EtdCyd are in the range reported for other 2'-deoxycytidine nucleosides (Young & Wilson, 1975; Kashino, Negishi & Hayatsu, 1988; Low, Tollin, Howie & Wilson, 1988; Sato, 1988; Silver-